

Plant Nutrients—Potassium

Information compiled by the California Fertilizer Foundation

Plant Utilization – Potassium, one of the 17 chemical elements required for plant growth and reproduction, is often referred to as the “the regulator” since it is involved with over 60 different enzyme systems in plants. Potassium helps plants to resist drought and effects from excessive temperatures. It also increases crop resistance to disease. Potassium aids plants in the production of starches, controls root growth, and regulates the opening and closing of pores in plant cells (called stomata), which is important for efficient water use.

All plants require potassium, especially crops high in carbohydrates, such as potatoes. Studies have shown that adequate amounts of potassium may promote the growth of long, strong cotton fibers; increase the shelf life of fruits; increase the stem length and quantity of roses; enhance the green color and growth of turf grass; and increase the size and quality of fruits, grains, and vegetables.

Production – Potassium is the seventh most abundant element in the Earth’s crust, yet only one to two percent is available to plants. The rest is fixed with other minerals and unavailable to plants. Farmers often apply potassium fertilizer for optimum plant growth.

Most potassium is mined from underground deposits. Most potassium is shaft mined, like coal. Some shafts are drilled as deep as 3,000 feet. In some cases, solution mining is also used in which case water is pumped into the shaft to dissolve the ore. The solution is extracted and allowed to evaporate, leaving behind potassium salts.

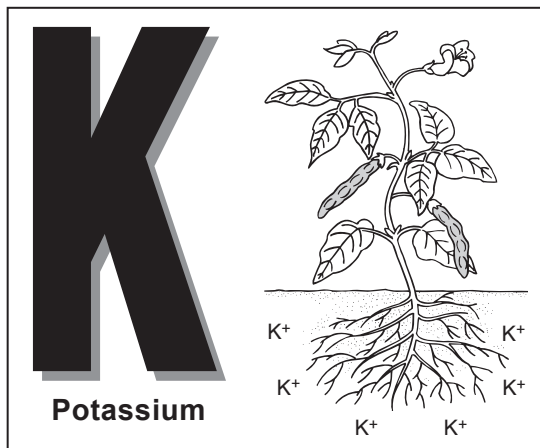
Some potassium comes from the evaporation of water from natural salt lakes, such as the Great Salt Lake in Utah, and the Dead Sea in Israel and Jordan. Tobacco stems, wool waste, sugar beet factory waste, and flue dust also contain potassium, but their use as a fertilizer is limited.

Forms – Potassium is symbolized as K_2O on fertilizer labels and is the third number on fertilizer labels. Plants absorb potassium in the form of the ion K^+ which dissolves readily in water.

Ninety-five percent of all potassium fertilizers come in the form of muriate of potash, also known as potassium chloride. For crops unable to tolerate chloride, potassium

sulfate, potassium nitrate, and other chloride-free salts are used. Potassium comes in both liquid and granular form and is usually mixed in the soil or placed directly near the root zones of plants. Application of chloride-free foliar sprays are sometimes used on certain crops.

History – The letter *K*, used to symbolize potassium, comes from the German word *kalium*. During Colonial times, people burned wood and other organic matter in pots to manufacture soap. The ashes were rinsed and the water was allowed to evaporate, leaving a residue of potassium salts. People called the residue “pot ashes” or potash. These salts were boiled with animal fat to produce soap.



In 1868, Samuel William Jackson, a botanist in Connecticut, burned plants and analyzed the ash. Jackson found plants consisted of large amounts of potassium, as well as other minerals. His work led to the use of fertilizers to promote an increase in crop yields.

Top Producing Regions – Canada leads the world potash production and exports, producing 8.4 million tons in 2007. Russia, Belarus and Germany are also top producers of potash. U.S. production has been stable with most domestic production occurring in New Mexico. Lesser amounts are produced in Utah and Michigan. The price of potash fertilizer has increased significantly in the past few years, causing mining companies to seek new sources of the raw material throughout the world.

China is the world’s leading potash consumer, using nearly 8 million tons in 2007. The U.S. and India are the next leading consumers of potash fertilizer. Approximately 20% of the 6.5 million tons of potash used in the U.S. is domestically produced.

Economic Value – United States farmers pay more than \$800 million annually for potassium fertilizers, with California farmers paying \$25 million for the product each year.

For additional information:

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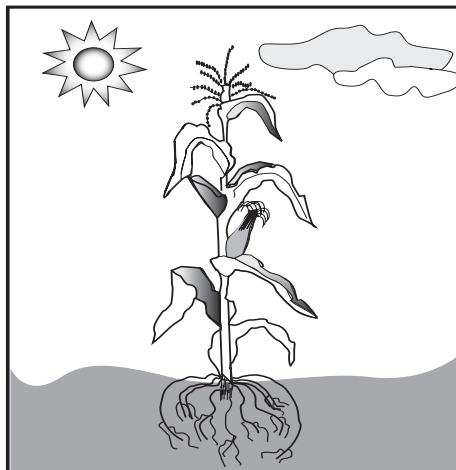
Web Site: www.calfertilizer.org



Potassium Activity Sheet

How Potassium Functions in Plants

- Helps retard crop diseases.
- Builds cellulose needed for stalk and stem strength.
- Aids in photosynthesis and food function.
- Increases root growth and improves drought resistance.



- Produces grain rich in starch.
- Necessary for plant protein formation.
- Reduces water loss and wilting.
- Assists many enzyme actions.

Lesson Ideas

- On a world map, color the major potassium exporters blue and the major importers red.
- On a map of North America, locate and color the areas where potassium is mined.
- Research how humans utilize potassium and find out what foods are high in potassium.
- Make a poster illustrating the various roles potassium plays in plant growth and health.
- Learn about Samuel William Jackson. Why was his research so important?
- Locate potassium on the Periodic Table of Elements. Learn about its physical and chemical properties.
- Find two points that are 3,000 feet apart so students can appreciate the depth of some potassium mine shafts.
- Research the Colonial soapmaking process and the various uses of potash.
- Find out how agronomists determine the potassium contents of soils.

Potassium Fun Facts

1. What country is the world's leading exporter of potassium?
2. How is most potassium obtained?
3. Why is potassium sometimes called "the regulator?"
4. Name two ways potassium helps plants.
5. Which state in the United States processes the most potassium?
6. Why is potassium called "potash?"
7. Is California a larger importer or exporter of potassium?
8. Name a large United States lake from which potassium is obtained.

1) Canada 2) Through underground mining 3) It regulates or controls many plant enzyme systems 4) It aides in protein and starch formation, stimulates root growth, provides winter hardiness, and opens and closes cell pores called stomata 5) New Mexico 6) Potash comes from pot ashes, residue found in wood ashes 7) Importer 8) The Great Salt Lake in Utah

Lesson Plan: The World of Potassium

Introduction: Potassium is an essential nutrient for plants and animals. It also has many other uses, depending on its chemical formulation. Your students will research potassium and its various uses in detail and create a wall-length mural, which depicts its importance.

Materials: Reference materials, including encyclopedias, human nutrition books, plant nutrient requirement books, butcher paper, paint or markers, glue.

Procedure:

1. Write the following phrases on index cards – plants which produce fibers for clothing; annual crops, such as celery; forage crops, such as alfalfa; tubers, such as potatoes; disinfectant; human nutrition; component in soap; plant guard cells; potassium forms which are usable by plants; agricultural by-products which contain potassium; roses and other flowers.

2. Divide the students into groups of three or four and distribute one index card to each group.
3. Each group is responsible for researching how potassium relates to the key words on the index card. After they gather the details, the group is to decide how they will depict their knowledge on a wall mural called "The World of Potassium."
4. In a class discussion, determine what the class mural will look like so that all aspects of potassium use will be displayed.
5. Have each group create their graphics and text for the mural and then place it on the mural.
6. Display the mural at a science night or in the library. This may be displayed with other murals made for other elements, such as nitrogen and phosphorus.

